

**Amendments to the Claims:**

The following listing of claims will replace all prior versions, and listings, of claims in the application:

1. (Currently Amended) An electrolytic phosphate chemical treatment method of forming a film composed of a phosphate compound and a metal that is reduced and precipitated from an ionic state on the surface of a metal material article to be treated by comprising the steps of performing the electrolytic treatment on said article ~~to be treated in a~~ phosphate chemical treatment bath by contacting said metal material article having electrical conductivity with said phosphate chemical treatment bath containing phosphate ions, ~~and~~ phosphoric acid, nitrate ions, metal ions that form a complex with the phosphate ions in said phosphate chemical treatment bath, and metal ions for which the dissolution-precipitation equilibrium potential at which the metal ions dissolved in said phosphate chemical treatment bath are reduced and precipitate as metal is equal to or greater than -830 mV, which is the cathodic reaction decomposition potential of ~~the a~~ solvent in the form of water when indicated as the hydrogen standard electrode potential, and is substantially free of metal ions other than those which are a component of the film; wherein,

the ~~ORP~~ (oxidation-reduction potential) (ORP) of said phosphate chemical treatment bath ~~(indicated as the potential relative to a standard hydrogen electrode)~~, is maintained at equal to or greater than 700 mV.

2. (Currently Amended) An electrolytic phosphate chemical treatment method according to claim 1, wherein said electrolytic treatment ~~preferably~~ uses for ~~the an~~ electrode material that dissolves in the treatment bath ~~a~~ the metal ions that forms ~~a~~ the complex with the phosphoric acid and the phosphate ions in the phosphate chemical treatment bath, ~~and/or~~ ~~a~~ the metal material for which the dissolution-precipitation equilibrium potential at which the ions dissolved in the phosphate chemical treatment bath are reduced and precipitate as the

metal is greater than or equal to -830 mV, which is the cathodic reaction decomposition potential of the solvent in the form of water when indicated as the hydrogen standard electrode potential, ~~and or~~ a metal material that is insoluble during the electrolysis.

3. (Currently Amended) The electrolytic phosphate chemical treatment method according to ~~either claim 1, wherein the~~ further comprising an amount of Fe ions dissolved into the treatment bath from an Fe electrode, when performing a cathodic treatment of said article to be treated and using ~~an the~~ Fe electrode as the electrode that dissolves in the treatment bath, is controlled in order to make said ORP of the phosphate chemical treatment bath equal to or greater than 700 mV.

4. (Currently Amended) The electrolytic phosphate chemical treatment method according to claim 1, wherein ~~in the a case where~~ the article to be treated is a steel material, the method further comprises the an amount of Fe ions dissolved into the treatment bath in an anodic treatment in which said steel material in the form of the article to be treated is dissolved as ~~the an~~ anode, and the amount of Fe ions that ~~dissolve~~ dissolves in the treatment bath from an Fe electrode in a cathodic treatment, are controlled so that the ORP of the phosphate chemical treatment bath is equal to or greater than 700 mV.

5. (Currently Amended) The electrolytic phosphate chemical treatment method according to claim 1, wherein ~~the an~~ electrode used in electrolysis for making the ORP of the phosphate chemical treatment bath equal to or greater than 700 mV is an insoluble metal material.

6. (Original) The electrolytic phosphate chemical treatment method according to claim 1, wherein a chemical that contains Fe ions which replenishes the phosphate chemical treatment bath is an Fe-phosphate complex in order to make the ORP of said phosphate chemical treatment bath equal to or greater than 700 mV.

7. (Original) The electrolytic phosphate chemical treatment method according to claim 1, wherein the ORP of the phosphate chemical treatment bath is equal to or greater than 770 mV.

8. (Currently Amended) The electrolytic phosphate chemical treatment method according to claim 1, wherein the metal ions that form ~~a~~the complex with the phosphoric acid and the phosphate ions in the phosphate chemical treatment bath are ~~preferably~~ at least one ~~type~~ of Zn, Fe, Mn or Ca ions.

9. (Currently Amended) The electrolytic phosphate chemical treatment method according to claim 1, wherein NO, NO<sub>2</sub> and/or N<sub>2</sub>O<sub>4</sub> gases generated and dissolved in ~~an electrolytic~~a treatment tank are removed from the treatment bath by separating the treatment tank into an electrolytic treatment tank that carries out the electrolytic treatment and an auxiliary tank that does not carry out electrolytic treatment, circulating the treatment bath between the two tanks, and providing a mechanism that opens ~~treatment liquid of the treatment bath~~ to the atmosphere either between the ~~above two tanks~~electrolytic treatment tank and the auxiliary tank or within the ~~two tanks~~electrolytic treatment tank and the auxiliary tank, as a means of separating the NO<sub>2</sub>, N<sub>2</sub>O<sub>4</sub> and/or NO ~~gas-gases~~ formed in the treatment bath accompanying the electrolytic treatment from the treatment bath.

10. (Currently Amended) The electrolytic phosphate chemical treatment method according to claim 9, wherein the auxiliary tank that does not carry out the electrolytic treatment has a mechanism in which the treatment liquid is passed through a permeable solid structure.

11. (Original) The electrolytic phosphate chemical treatment method according to claim 10, wherein the solid structure is a film.

12. (Currently Amended) The electrolytic phosphate chemical treatment method according to claim 9, wherein a filter having a mechanism that filters the treatment liquid is used for the auxiliary tank that does not carry out the electrolytic treatment.

13. (Currently Amended) The electrolytic phosphate chemical treatment method according to claim 9, ~~having a liquid circulation circuit that removes~~ further comprising removing, through a liquid circulation circuit, a portion of the treatment liquid at a location prior to being introduced into a filter material in a filter, ~~exposes~~ exposing the removed treatment liquid to the atmosphere, and ~~returns~~ returning it to the electrolytic treatment tank after removing the gases in the form of nitrogen oxides present in the treatment liquid.

14. (Currently Amended) The electrolytic phosphate chemical treatment method according to claim 9, wherein ~~the ORP~~ an oxidation-reduction potential of the treatment bath is equal to or greater than ~~840~~ 840 mV.

15. (Currently Amended) The electrolytic phosphate chemical treatment method according to claim 9, wherein the treatment bath is maintained in a constant state by measuring ~~the above ORP~~ an oxidation-reduction potential value of the treatment bath and changing ~~the~~ an amount and/or composition of replenishing chemical corresponding to the change in that value.